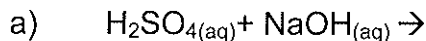


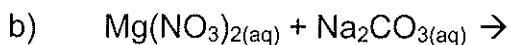
Writing Ionic Equations

Complete the formula equation and write the complete ionic and net ionic equations for each.



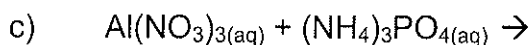
Complete:

Net:



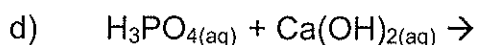
Complete:

Net:



Complete:

Net:

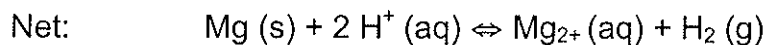
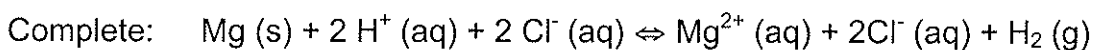
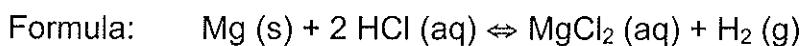


Complete:

Net:

Write chemical equations, total ionic equations and net ionic equations for each reaction. The first one is done for you. (Assume that all reactions occur)

Magnesium metal is placed in hydrochloric acid



Zinc metal is placed in silver nitrate solution

Formula:

Complete:

Net:

Barium chloride solution is added to lead (II) nitrate solution.

Formula:

Complete:

Net:

Sulphuric acid is added to calcium hydroxide solution.

Formula:

Complete:

Net:

Sodium carbonate solution is added to nickel (III) nitrate solution.

Formula:

Complete:

Net:

Aqueous chlorine is added to sodium bromide solution.

Formula:

Complete:

Net:

Nitric acid is added to aluminum hydroxide solution.

Formula:

Complete:

Net:

Sodium hydroxide is neutralized with oxalic acid.

Formula:

Complete:

Net: